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(54) Title: METHOD FOR PREPARING PIPERAZINES

(57) Abstract

The invention provides a method of synthesizing N-substituted 1,4-piperazines and N-substituted 1,4-piperazinediones. The invention also provides a method for synthesizing combinatorial libraries of piperazine derivatives which can be assayed for biological activity in prespecified ligand binding or enzymatic activity screens.

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METHOD FOR PREPARING PIPERAZINES

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Field of the Invention

This invention relates to a method of generating large numbers of diverse non-peptide organic compounds which can be assayed for biological activity in pre-specified ligand binding or enzymatic activity screens. More specifically the diverse organic compounds formed are N-substituted 1,4-piperazines or their corresponding piperazinediones formed by "on-resin" or solid-phase cyclization of unnatural N-substituted dipeptoids. These compounds are useful in that they serve as an inexpensive source of chemical diversity to identify low molecular weight lead compounds with desired ligand affinity or enzyme inhibitory activity, which can be further optimized by combinatorial or traditional medicinal chemical methodology.

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Background of the Invention

Traditionally the most effective strategies for identifying novel non-peptide lead structures which have affinity for a targeted receptor or enzyme, which normally binds peptidic ligands, have included mass screening of diverse compounds. Mixtures of natural products, usually obtained as extracts, or chemical files of synthetic compounds are traditionally the source of diverse chemical structures for lead identification. Often the lead compound may possess relative weak affinity for the targeted receptor or enzyme; the affinity can then be increased by systematic, empirical modification of the lead structure.

Lead compounds are found by screening of natural products obtained from plant, soil, or animal sources, or fermentation with microorganisms in diverse media. Synthetic compounds contained in historical collections of pharmaceutically interesting molecules or their synthetic precursors are also screened. Both natural and synthetic sources of chemical diversity present their own unique problems and limitations. Unfortunately natural product leads are often extremely difficult to identify, purify, and

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synthesize. Increasingly the use of natural products leads creates ethical dilemmas in that the organism which produces the lead compound may be rare and the discovery of a new source of biological activity may endanger the natural specie which produces the chemical, or the biome where the specie is found. Historical chemical libraries collections produced by pharmaceutical companies are often limited in diversity and scope since the original compounds were originally designed to target specific diseases or pathophysiological mediators, and more importantly diverse collections chemical files are unavailable to researchers who do not possess enormous financial resources.

Synthetic combinatorial libraries of diverse peptides have only partially answered the need for new methods to identify novel ligands for pharmaceutically interesting targets. Peptides suffer from a number of disadvantages including poor oral bioavailability, poor in vivo stability and high cost. Although a very large number of diverse peptides can be made, their structural diversity is limited to natural or readily synthesible amino acid building blocks and their linear polymeric character. The transformation of a peptide lead into an orally available, biologically stable, clinically useful drug, is an extremely time consuming and expensive research project and has met with limited success.

Recently several solid phase combinatorial approaches have been developed for creating diverse libraries based on small scaffolds which have been previously incorporated in successful drugs. These scaffolds include heterocycles such as benzodiazepines (Bunin et al., J. Am. Chem.

25 Soc. 114 10997 (1992)), hydantoins (DeWitt, Proc. Natl. Acad. Sci. USA 90 6909 (1993)) and piperazinediones (Gordon et al., Biorganic and Medicinal Chemistry Letters, 5 47 (1995)) and non-heterocyclic scaffolds such as biphenyls (Pavia et al, PCT/US94/07780 (WO 95/04277). The present invention provides a method for producing 1,4-piperazine-2,5-diones or their corresponding reduced 1,4-piperazine analogs with greater potential for diversity in structure and with a chemical synthesis which offers advantages in yield, purity, cost and ease of production over known methods of producing diverse piperazinediones.

Piperazines and piperazinediones are common scaffolding 35 structures found in a wide variety of bio-active compounds and drugs. Examples of this disclosed invention have been found to bind to human

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bradykinin, human neurokinin 1 and 2, human mu and kappa (animal) opioid receptors, and are expected to function as antagonists in many G-protein coupled receptors or inhibitors of many classes of enzymes.

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Detailed description of the invention

The present invention comprises a method for producing large numbers of structurally diverse piperazines and piperazindiones of the general structures illustrated below.

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Where R¹ through R² are independently selected from

hydrogen or C1 to C12 alkyl, C3 to C8 cycloalkyl, or alkylaryl pendant
groups. The pendant groups may be substituted with guanidino,
alkylguanidino, benzamidino, methoxy, ethoxy, methyl, ethyl, propyl,
butyl, piperidinyl, piperazinyl, indanyl, indolyl, quinuclidinyl, hydroxyl,
amino, alkylamino, aminoalkyl, biphenyl, napthyl, imidazoyl, pyridinyl,
purinyl, benzoxazolyl, carboxamides, sulfonamides, carboxylates, pyrazoles,
pyrimidines, fluoro, chloro, bromo, or trifluromethyl moeities.

In a preferred embodiment of the invention R¹ is selected from cyclohexyl, methylcyclohexyl, benzyl, substituted benzyl, phenethyl, biphenyl, benzoyl, hydrogen, methylnapthyl, indanyl, guanidinoalkyl (C2 to C8), aminoalkyl (C2 to C8), arylalkylcarbamates with alkyl chains of C2 to C8, alkanoyl (C1 to C12), and alkyl (C1 to C12) or hydrogen.

In a preferred embodiment of the invention R² is selected from benzyl, substituted benzyl, indole, methylnapthyl, cyclohexyl, methylcyclohexyl, indanyl, (C1 to C5)alkyloxyaryl, aminoalkyl (C1 to C8), guanidino alkyl (C1 to C8), alkylcarboxylic,hydroxyalkyl(C1 to C5), alkylthioalkyl, (C1 to C3) alkylbenzamidino, or hydrogen.

In a preferred embodiment of the invention R³ is selected from is selected from cyclohexyl, methylcyclohexyl, benzyl, substituted benzyl,

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phenethyl, biphenyl, benzoyl, hydrogen, methylnapthyl, indanyl, guanidinoalkyl (C2 to C8), aminoalkyl (C2 to C8), arylalkylcarbamates with alkyl chains of C2 to C8, alkanoyl (C1 to C12), alkyl (C1 to C12), (C2 to C12)-alkylguanidinoalkyl(-C2 to C8), benzamidino, (C1 to C6) alkylbenzamidino, (C1 to C6)alkylquinuclidine, (C1 to C6) alkyl piperazine, C1 to C6 alkyl N-substituted piperazine, (C1 to C6) alkylamino piperidine, (C1 to C6) alkylamino N-substituted piperidine (C1 to C6) alkylcarboxamidoquinuclidine, (C1 to C6) alkylcarboxamidobenzamidino, (C1 to C6)alkylcarboxamidoquinuclidine, (C1 to C6) alkylcarboxamido piperazine, C1 to C6 alkylcarboxyl N-substituted piperazine, (C1 to C6) alkylcarboxamidino piperidine, or (C1 to C6) alkylcarboxamidino N-substituted piperidine, or hydrogen.

In another preferred embodiment R⁴ and R⁷ are independently defined as for R1; and R⁵ and R⁶ are independently defined as for R² or R³. In this disclosure two most preferred methods for synthesizing independent purified piperazinediones and piperazines are described. A preferred method for preparing a diverse combinatorial library of piperazinediones and piperazines with several thousand discrete components from which lead bradykinin antagonist structures and other bio-active compounds with G-protein binding properties have been discovered is also disclosed.

The compounds or libraries of compounds of this invention are preferably synthesized as described in schemes I and II below. The compounds of the invention could also be synthesized by those skilled in the art by adaptation of the solid phase methodology described here to solution phase methods; such an adaptation would be considered an extension of the current invention.

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A suitably N-protected α-amino acid is attached to a polymeric support using methods well known in the art. These methods include displacement of benzylic halogens to form ester linkages utilizing cesium carbonate, potassium carbonate or potassium iodide to accelerate the reaction. Ester formation can also be accomplished by the action of carbodiimide coupling reagents such as dicylohexylcarbodiimide or diisopropylcarbodiimide and a benzylic alcohol, utilizing tertirary base, with or without 4-dimethylaminopyridine as catalyst. Reagents such as BOP, HBTU, and BOP-Cl can also effectively be used to derivatize the resin. An advantage of the current invention is that the resin need not be a

benzylic ester support such as Merrifield or Wang type resins, since acidolysis is not required to free the piperazinedione from the resin. The solid support may be an alkyl derivative such as polyethylene glycol or any other polymeric or discrete hydroxyl group. A very large number of suitably N-protected resin-immobilized α-amino acids are commercially available and all but the most acid sensitive of these constructs are applicable in the current invention.

Scheme I

cyclizations, etc.

In method A, illustrated in scheme I, the resin immobilized amino acid is deprotected according to methods well known in the art. These methods may include treatment with trifluoroacetic acid for Bocprotected amino acids, treatment with secondary amines such as piperidine for Fmoc-protected amino acids or treatment with hydrogen or organicpalladium complexes for reducible protecting groups. The resulting free amino group is then reductively alkylated with a suitable adehyde, 10 preferably by treatment with NaCNBH3 in the presence of catalytic acetic acid in dimethylformamide. A number of alternative methods for accomplishing reductive amination are known in the art and include hydrogenation in the presence of soluble organic substituted metal catalysts, variously substituted borohydride reagents or the use of Lewis Acids such as Ti(O-Pr)4 and sodium cyanoborohydride. The resulting secondary amine is facilely bromoacetylated with bromoacetic acid and diisopropylcarbodiimide or dicyclohexylcarbodiimide. This reaction can also be accomplished with other coupling reagents such as HBTU. The resulting bromoacetyl compound is then reacted with concentrated primary 20 amines in a polar aprotic solvent such as DMSO to provide an N-substituted

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glycyl residue. The key step to the synthesis in this disclosure is the solidphase mediated simultaneous cyclization and cleavage of the incipient dipeptoid from the solid phase resin. This is accomplished by heating in a non-reactive solvent such as 2-butanol in the presence of a catalyst such as acetic acid. The boiling point, polarity, solvent characteristics and hydroxylic character of racemic 2-butanol all contribute to the success of the cyclization method. The simultaneous cyclization and release of piperazinediones has numerous advantages:

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1. Since compounds which are not dipeptoids are not released from the resin by these mild conditions, failure sequences or potential impurities are not freed from the resin to contaminate the desired piperazinediones. Crude purities, following aqueous work-up, in excess of 95% can be obtained in selected cases.

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2. The mild deprotection chemistry involved allows a very large number of acid sensitive substituents and acid sensitive protecting groups to be employed, we have routinely produced diketopiperizines containing versatile protecting groups such as the Boc and Cbz group which were then removed for further modification of the molecule.

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3. The reaction scheme is simple and provides compounds of superior purity without resorting to expensive and sensitive reagents such as PyBrop or its relatives or corrosive acids such as TFA; such considerations increase the propensity for the application of this method to automated reaction devices.

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4. Since the cyclization/cleavage reaction is driven by high temperature and is a highly favored intramolecular formation of a piperazindione, amide bonds can be formed which are very difficult to form by alternative methods. For example the solution and solid phase formation of the dipeptoid illustrated was attempted using a number of reagents including PyBrop, the reagent of choice for the synthesis of peptoids involving couplings at hindered secondary amines and suggested for the solid-phase synthesis of

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piperazinedione libraries, (D. W. Gordon Biorganic and Medicinal Chemistry Letters, 5 47 (1995)) in our hands we obtained essentially none of the desired compound. Utilizing the methodology of this invention (method B) a corresponding difficult amide bond was formed very readily during the cleavage step. Consequently we believe that piperazindiones are readily available utilizing this methodology which can not be easily formed by other reported methods which involve couplings of very hindered secondary amines.

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An additional advantage is accrued by the addition of the R³ substituent by displacement with a primary amines via method A. The introduction of complex groups via primary amines greatly increases the diversity of substituents for the side chain over methods which only employ aldehydes as in reductive alkylation reactions. The supply of commercial primary amines is much more diverse than less stable commercial aldehydes. A very large number of suitable primary amines are available commercially or through simple synthetic steps making an enormous number of diverse compounds possible. The geometric orientation of such

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N-substituted species ffers increased diversity over piperazinediones obtained from simple dipeptides.

In method B a dipeptide may be synthesized on a supporting resin by methods well known in the art, deprotection of the dipeptide

5 produces an N-terminal amine which is reductively alkylated as described in method A. Again the key step to the synthesis in this invention is the solid-phase mediated simultaneous cyclization and cleavage of the incipient dipeptoid from the solid phase resin. This is accomplished by heating at reflux in a non-reactive solvent such as 2-butanol in the presence of a

10 catalyst such as acetic acid. As in method A a large number of additional reactions and deprotections can subsequently be carried out on the free diketopiperazines. We illustrate a number of deprotection reactions, acylations and guanidination reactions.

Piperazinediones produced by either method A or method B can be reduced to the corresponding piperazine structure by treatment with strong reducing agents such as diborane or lithium aluminum hydride.

In many instances the compounds that are produced are of high purity and may be assayed directly or they may be purified by common chromatographic techniques such as preparative reverse phase high performance chromatography or ion exchange chromatography.

In order to produce large numbers of diverse analogs, mixtures of precursor reagents may be formed which produce mixtures of compounds which differ in structure in the various side chain substitutions. A simple method for producing mixtures of controlled content is to 25 derivatize individual resin samples with a given amino acid substituent and then mix portions of such resins, each portion derivitized with a unique amino acid substituent. The resins are then split into aliquots and reacted with different reagents which produce a variety of substituents at one given position. Each aliquot is then split into new aliquots which are reacted with 30 reagents which provide a variety of substituents at a new position. In the examples of this patent we illustrate this process by providing a resin mixture which is derivatized by 10 different amino acid derivatives. (This number could be much larger, but a simple example is chosen for clarity.) This resin is split into 12 aliquots. Each of 10 aliquots is reductively alkylated with a unique aldehyde. Each resulting aldehyde is split after bromoacetylation into 12 new aliquots, each of which is reacted with a

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unique primary amine. The resulting compound is then cyclized and each resulting mixture is split in half, half of the aliquot is then reduced with lithium aluminum hydride to provide piperazines. Two of the original aliquots are not reacted with aldehydes but are bromoacetylated, reacted with amines, cyclized, reduced to piperazines first and then acylated with either benzoic acid or dodecanoic acid. The various processes illustrated here produce a large number of mixtures of compounds, where each mixture contains at least 10 unique piperazines or piperazindiones. These mixtures are conveniently assayed by HPLC and MALDI -Mass

10 Spectrometry, and then tested after dissolution in DMSO for activity in selected biological assays. We illustrate broad screening on a human bradykinin receptor clones. Human neurokinin type I and II receptor clones, human mu and guinea pig brain kappa opioid receptor clones. Compounds can be selected for specific identification and resynthesis on the

15 basis of an apparent level of inhibition at a given concentration.

This process is concisely illustrated by the discovery of two bradykinin human receptor ligands as potentially pharmaceutically useful leads for the development of potent bradykinin antagonists. Screening of example combinatorial mixture 219 at an average estimated concentration of 20 10 uMolar produced 77% inhibition of radio-labeled bradykinin binding in the human receptor clone assay. This sample was subjected to a 5-60% CH3CN, 30 minute, then 60-90%, 15 minute gradient in water containing constant 0.1% TFA on a 25 mm reverse phase HPLC column, (10 mL/min flow rate). Fractions of approximately 10 mL were collected. Aliquots of 25 the collected fractions were rescreened for BK binding inhibition and fraction 34 exhibited 65% inhibition. The active component of fraction 34 was identified with MALDI Mass spectrometry as example 60. This compound was resynthesized in pure form according to method A, and assayed for BK binding inhibition, an IC50 value of 4.1 uMolar (human BK2 30 receptor) was obtained. In an analogous fashion, combinatorial mixture example 166, exhibited 71% inhibition at an estimated average concentration of 10 uMolar. The sample was submitted to chromatography, binding assays of fractions, mass-spectrometry and the example 61 was identified as the active component and synthesized in pure form by method 35 A. An IC₅₀ value of 13 uMolar (human BK₂ receptor) was obtained.

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Numerous other human receptor ligands have been identified from this single combinatorial library set.

Examples

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Example 1. General Procedure for Resin Derivitization:

In a solid phase peptide synthesis device (Stewart et al., Solid Phase Peptide Synthesis, Pierce Chemical Company, 1984), pre-silanized with Sigmacote, Wang peptide synthesis resin (p-benzyloxybenzyl alcohol [copolystyrene-1%-divinyl benzene] resin, 0.2 to 0.80 meqv. / g), 300 mg to 10 g is suspendend in DMF (10 volumes) and then the solvent is filtered away. The resin is washed two additional times with DMF. The resin is washed three times with dichloromethane, then three times with DMF. Finally the resin is suspended in DMF and allowed to mix with nitrogen bubbling for 30 to 60 minutes. In a separate flask 5 to 10 equivalents of N^{α} Fmoc-Protected amino acid is dissolved in the minimum volume of dry DMF and treated with 0.2 equivalents 4-dimethylaminopyridine, and 10 equivalents of N,N'-diisopropylcarbodiimide. The reaction is transferred to the synthesis reaction vessel containing the Fmoc-protected Amino Acid resin and the mixture is agitated for 15 hours with nitrogen bubbling. The reaction mixture is filtered away and the resin is washed sequentially, with DMF, methanol, DMF, and dichloromethane, the resin is dried in a stream of dry nitrogen, then under high-vacuum. The amount of Fmoc-amino acid added to the resin is estimated colorimetrically as follows: A 5 mg sample of the resin was suspended in 6.0 mL of DMF containing 20% piperidine. After 10 minutes of gentle agitation, the solution is decanted and the absorbance measured at 290 nM. The measured absorbance at 290 nM is multiplied by 0.2424 to yield the resin substitution in units of miliequivalents per gram of derivatized resin.

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Example 2. General Procedure for Synthesis of Piperazinediones According to Method B.

A. Resin Deprotection

The resin immobilized N α -Fmoc-amino acid (side-chain R⁵) is deprotected by reaction with 25% piperidine in DMF (\sim 20 mL / g resin)

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with agitation by nitrogen bubbling for 25 minutes. The solution is filtered away and the resin is reacted again with 25% piperidine in DMF (\sim 20 mL / g resin) with agitation by nitrogen bubbling for 25 minutes. The solution is filtered away and the resin is washed well with DMF.

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B. Amino Acid Activation

Symmetrical Anhydride Method: In a separate flask 4 equivalents of the Fmoc-amino acid (side-chain R⁶) are dissolved with warming in the minimum volume of dichoromethane containing 10% DMF, the solution is cooled to room temperature and 2 equivalents of dicyclohexylcarbodiimide are added. The reaction is stirred for 30 minutes at room temperature; the resultant dicyclohexylurea (DCU) is filtered away and the filtrate is added to the resin prepared above. The reaction mixture is agitated for 2 hours with nitrogen bubbling.

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Benzotriazole Active Ester Method:

Difficult couplings such as those employing Fmoc-lgl-OH can be accomplished by preforming the active ester by the reaction of four equivalents of the Fmoc-amino acid in DMF with four equivalents of hydroxybenzotriazole and dicyclohexyl carbodiimide. After 30 minutes (DCU) is filtered away and the reaction mixture is added to the resin as above. Similar preactivation with HBTU/HOBt and 8 equivalents disopropylethyl-amine is acceptable for very difficult couplings.

25 C. Resin Deprotection:

As described in step A.

D. Reductive Amination

Dried resin is suspended in DMF and agitated for 30 minutes; the DMF is filtered away and the resin is then treated with DMF (~ 40 mL/mmole) containing 1% acetic acid. Three equivalents of the appropriate aldehyde is added followed by 10 equivalents of sodium cyanoborohydride. The reaction is agitated by nitrogen bubbling for 6 to 24 hours. Reaction progress can be accessed by the Kaiser test (Ibid pg 76) or by HPLC analysis of a TFA cleavage reaction of a few resin beads. If incomplete reaction is obtained the reaction is filtered and resubmitted to the reductive amination

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conditions. When complete reductive amination is achieved the reaction mixture is filtered and washed well with DMF then dichloromethane, followed by another series of DMF washes. The resin may be dried if storage is required after the dichloromethane washes.

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E. Cyclization:

Dried resin or slurried aliquots are transferred to a round bottom flask or mini-reactor tube and treated with 2-butanol containing 11% (v/v) acetic acid (~ 5 to 10 mL/mg resin). The reaction vessel is placed in a 110 ° oil bath and heated at reflux under nitrogen for 48 to 72 hours. The crude piperazinedione is isolated by filtration, and washing of the resin with acetonitrile, methanol, and dichloromethane. The volatile solvents are removed by rotary evaporation. The product may be purified by HPLC chromatography. Side chain protecting groups such as Cbz and Boc are conveniently deprotected at this stage, before LiAlH4 reduction. Free amino groups formed by deprotection may be acylated or guanidinated.

20 Example 3. General Procedure for Guanidination of Side-Chain Amino Groups Derived from N-Boc-Protected Moieties.

The crude diketopiperazine obtained by removal of the 2-butanol/acetic acid solution is dissolved with excess (>15 equivalents) of trifluroacetic acid as a 50% v/v solution in dichloromethane. The reaction is stirred one hour to insure removal of the side chain N-Boc protecting group. The solvent is then removed on the roto-vap. The residue is treated with dichloromethane and the solvent is again removed using rotary evaporation. The sample is again treated with dichloromethane and the solvent removed using rotary evaporation. The residues are dissolved in DMF (0.1 to 1 mL/20 mg crude product) and treated with two equivalents of pyrazole-1-carboxamidine hydrochloride (Bernatowicz et al., J. Org. Chem. 57 2497 (1992)) and 5 equivalents of N,N-diisopropyl-ethylamine. The reaction is stirred 6 to 24 hours and then is treated with excess TFA in dichloromethane. The solvent is evaporated and the crude product is precipitated with anhydrous diethylether and purified by reverse phase HPLC chromatography. Primary amino groups in piperazines produced by

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reduction of piperazinediones by LiAlH4 are guanidinated using the identical procedure in DMF with two equivalents of pyrazole-1-carboxamidine hydrochloride and 5 equivalents of N,N-diisopropylethylamine. The reaction is stirred 6 to 24 hours and then treated with excess TFA in dichloromethane. The solvent is evaporated and the crude product is precipitated with anhydrous diethylether and purified by reverse phase HPLC chromatography.

10 Example 4. General Procedure for Guanidination of Side-Chain Amino Groups Derived from N-Cbz-Protected Moieties.

The crude diketopiperazine obtained by removal of the 2-butanol/acetic acid solution is suspended in acetic acid (~0.5 to 1 mL / 100 mg crude product) and treated with excess (>15 equivalents) of

15 hydrobromic acid in a 30% solution in acetic acid. The reaction is stirred for 30 minutes to insure removal of the side chain N-Cbz protecting group. The crude product is precipitated by addition of anhydrous diethylether and collected by centrifugation. The primary amine (as the HBr salt) is then guanidinated as described in example 4. In general compounds are

20 purified by reverse phase HPLC chromatography although crude puritys of 95% have been obtained.

Example 5. General Procedure for LiAlH₄ Reduction of Piperazinediones from Method A to Piperazines.

The crude piperazinedione after cyclization or side-chain deprotection is dissolved under nitrogen with 20 equivalents of a 1 M LiAlH₄ solution; the solution is then heated at reflux under nitrogen for 24 hours. The reaction is quenched with ice and the volatiles removed by rotary evaporation. The residue is extracted with acetonitrile, filtered, acidified with TFA to ~pH 3, and the product purified by reverse phase HPLC or Sep-Pak filtration using a step gradient of 0-50% acetonitrile.

35 Example 6. General Procedure for LiAlH₄ Reduction of Piperazinediones from Method B to Piperazines.

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The crude piperazinedione after cyclization or side-chain deprotection is dissolved under nitrogen with 20 equivalents of a 1 M LiAlH₄ solution; the solution is then heated at reflux under nitrogen for 24 hours. The reaction is quenched with ice and the volatiles removed by rotary evaporation. The residue is extracted with acetonitrile, filtered, acidified with TFA to ~pH 3, and the product purified by reverse phase HPLC or Sep-Pak filtration using a step gradient of 0-50% acetonitrile.

10 Example 7. General Procedure for Acylation of Piperazines.

Piperazines containing primary or secondary amines are dissolved in DMF (0.05 to .5 molar) and treated with 1.25 equivalents of the appropiate carboxylic acid, 1.25 equivalents of diisopropyl-ethylamine and 1.25 equivalents of HBTU. The reactions are agitated, stirred or sonicated for 2 to 16 hours in polypropylene conical capped tubes. Each tube is treated with a mixture of 1 N NaOH and EtOAc. After vigorous mixing, the aqueous layer is removed by suction through a capillary, and the NaOH wash repeated. The organic layer is evaporated utilizing a speed-vac (high speed centrifugal vacuum evaporator) and the samples are purified by HPLC or step- wise gradient (0 - 80% acetonitrile in water) on a Sep-Pak column. Samples are evaporated utilizing the speed-vac, and dissolved in DMSO for biological testing.

25 Example 9. General Method for Synthesis of Piperzinones According to Method A.

The N°-Fmoc- amino acid derivitized resin is treated with DMF in a silanized (Sigmacote) reaction vessel, and agitated for 30 minutes. The DMF is filtered away and the resin is treated with 25 % piperidine in DMF (~20 mL / g resin). The deprotection reaction is agitated with nitrogen for 25 minutes. The solvent is then filtered away, and the resin reacted again for 25 minutes with 25% piperidine in DMF with nitrogen bubbling. The reaction mixture is filtered away and the resin is washed well with DMF, dichloromethane, methanol, and dichloromethane. The resin can then be stored under nitrogen at low temperature. Dried resin is suspended in DMF and agitated for 30 minutes; the DMF is filtered away and the resin is

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then treated with DMF (~ 40 mL /mmole) containing 1% acetic acid. Three equivalents of the appropriate aldehyde is added followed by 10 equivalents of sodium cyanoborohydride. The reaction is agitated by nitrogen bubbling for 6 to 24 hours. Reaction progress can be assessed by the Kaiser test or by 5 HPLC analysis of a TFA cleavage reaction of a few resin beads. If incomplete reaction is obtained, the reaction is filtered and resubmitted to the reductive amination conditions. When complete reductive amination is achieved the reaction mixture is filtered and washed well with DMF then dichloromethane, followed by another series of DMF washes. The resin immobilized N-substituted amino acid is then bromoacetylated with a solution of 12 equivalents of bromoacetic acid and 13.2 equivalents of diisopropyl-carbodiimide in DMF (~ 10mL /g resin). The reaction is agitated for 30 minutes, filtered, washed with DMF, and treated a second time with 12 equivalents of bromoacetic acid and 13.2 equivalents of diisopropyl-carbodiimide in DMF (~ 10mL /g resin). After 30 minutes of nitrogen agitation the solution is filtered, washed well with DMF, dichloromethane and dried under a nitrogen stream. The resin is then dried under vacuum. The resin is treated with a solution of an appropriate primary amine (0.25 to 2 M) in DMSO (~1 mL / 100 mg of resin). The 20 reaction may be sonicated for 3 hours or agitated for approximately 15 hours. The mixture is then filtered, washed well with DMF, dichloromethane, methanol, and finally with dichloromethane. The derivatized resin is then dried in a stream of nitrogen and then under vacuum if storage is necessary. The dried resin or slurried aliquots are 25 transferred to a round bottom flask or mini-reactor tube and treated with 2butanol containing 11% (v/v) acetic acid (~ 5 to 10 mL / mg resin). The reaction vessel is placed in a 110 o oil bath and heated at reflux under nitrogen for 48 to 72 hours. The crude piperazinedione is isolated by filtration, and washing of the resin with acetonitrile, methanol, and 30 dichloromethane. The volatile solvents are removed by rotary evaporation. The product may be purified by HPLC chromatography. Side chain protecting groups such as Cbz and Boc are conviently deprotected at this stage, before LiAlH4 reduction. Free amino groups formed by deprotection may be acylated or guanidated.

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Example 10. General Method for Combinatorial Synthesis of Mixtures of Piperzin nes According t Meth d A.

A large combinatorial library of piperazines and piperazinediones was synthesized as follows.

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A. Resin Mixing, Deprotection and Slurry Splitting:

The following $N^{\alpha_{-}}$ Fmoc amino acid resins were mixed in a large silanized solid phase peptide synthesis apparatus.

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Fmoc-Amino Acid	Loading (meq/g)	mMoles	Amount used.
L-Phenylalanine	0.72	0.625	0.87 g
D-Phenylalanine	0.81	0.625	0.77 g
L-Tryptophan	0.66	0.625	0.95 g
D-Tryptophan	0.57	0.625	1.10 g
D,L-Cha	0.57	1.25	2.20 g
D,L - 2-Nal	0.55	1.25	2.28 g
D.L-Igl	0.50	1.25	2.50 g
10 (5 D,L Pairs)	0.625 Average	6.25 Total	10.67 g total
Total	Ū		J

The resins were suspended in DMF and agitated for 30 minutes, the DMF was filtered away and treated with 200 mL of 25%

15 piperidine in DMF with nitrogen agitation for 25 minutes. The solution was drained away and the resin was again treated with 200 mL of 25% piperidine in DMF with nitrogen agitation for 25 minutes. The solution was filtered away and the resin was washed well with DMF and dichloromethane. The resin was slurried in a mixture of 3:2

20 dichloroethane:dimethylformamide (V/V), and the the slurry was volumetrically divided into 12 equal portions which were filtered, washed with dimethylformamide and dichloromethane and dried under vacuum. These 12 resin lots were given letter designations A through L for simplification for experimental description.

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B: Reductive Amination

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Each resin lot was reacted with the appropriate aldehyde or

ketone as follows.

Α	cyclohexanone
В	cyclohexane carboxaldehyde
С	benzaldehyde
D	phenylacetaldehyde
E	4'-biphenylcarboxaldehyde
F	No reductive amination.
G	2-Napthaldehyde
H	2-indanone
I	2-(N-Cbz)-ethanal
J	6-(N-Cbz)-hexanal
K	12-(N-Cbz)-dodecanal

5

Each resin lot was placed in a 50 mL solid phase reaction vessel, suspended in dimethylformamide (20 mL) and agitated for 30 minutes. The solvent was drained away and the resin was treated with 25 to 30 mL of dimethylformamide containing 3 equivalents (1.56 mmole) of aldehyde or ketone and 10 equivalents (327 mg) of sodiumcyanoborohydride. The reaction was agitated for 6 to 24 hours, and monitored by the Kaiser test. If incomplete reaction had occured the mixture was filtered away and a fresh solution of 25 to 30 mL of dimethylformamide containing 3 equivalents (1.56 mmole) of aldehyde or ketone and 10 equivalents (327 mg) of sodiumcyanoborohydride was added and reaction was continued for 6 to 24 hours. The reaction mixture was then filtered away and the resin mixtures were washed well with dimethylformamide and dichloromethane and dried under vacuum.

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C. Bromoacetylation:

All lots A through L were bromoacetylated as follows. Each resin sample was placed in a 50 mL solid phase synthesis reaction vessel and suspended in DMF. The mixture was agitated for 30 minutes, and filtered away. The resin was then reacted with 12 equivalents bromoacetic

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acid (867 mg) and 13.2 equivalents of diisopropylcarbodiimide (1.08 mL) in DMF (~ 10 mL /g resin). The reaction was agitated for 60 minutes, after which the solution was filtered away. The reaction vessel was charged with fresh bromoacetylation cocktail and the reaction was again agitated for 60 minutes. The solution was drained away and each lot was washed well with dimethylformamide followed by dichloromethane and dried under vacuum.

D. Amine Displacement Reactions

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Amines #1 through #12 used in displacement reactions:

D. Amine Displacement Reactions

Each lettered lot of Resin (A through L) following bromoacetylation was again subdivided into 12 sub-lots and numbered sequentially i. e. (A1 through A12, B1 through B12, ... L1 through L12). Each sublot was reacted with the corresponding numbered amine as follows.

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Amine#	<u>Equivalents</u>	Concentration in DMSO
1	20	0.43 M
2	20	0.43 M
3	9	0.193 M
4	13	0.284 M
5	14.5	0.315 M
6	14	0.302 M
7	20	0.43 M

	21	
8	23	0.50 M
9	23	0.50 M
10	23	0.50 M
11	17.5	0.38 M
12	20	0.43 M

Reactions were carried out in a specially designed multiple reaction device which is described in example 438. The 5 ml reaction vessles were pre-loaded with the appropriate amine solution in DMSO. Hydrochloride salts were treated with 2 equivalents of diisopropylethylamine in situ to provide free amino groups for the displacement reaction. The reactions were allowed to swirl for 15 hours or were sonicated at approximately 50 °C for 3 hours. The reaction flasks were then removed and the resin samples were washed by positive N₂ pressure using DMSO, DMF, dichloromethane, methanol, and dichloromethane.

15 E. Cyclization Reactions:

A fresh set of 12 5 mL reaction vessels charged with 3 mL of 2-butanol containing 2 M acetic acid are connected to the multiple reaction device so each resin sample is immersed in solution. The exposed gas dispersion tubes are connected to positive pressure nitrogen lines and cooled externally with dry ice. The 5 mL reaction vessels are placed in a 110 °C oil bath and heated for 48 to 90 hours. From time to time fresh butanol containing 2M acetic acid is added to maintain approximately 3 mL volume. After the reflux period the reaction flasks containing the now cleaved and cyclized piperzazine-diones are removed and the solvent removed by rotary evaporation. Completed piperazinedione mixtures are assayed by HPLC and laser desorption mass spectrometry, dissolved in a predetermined volume of DMSO and assayed for receptor binding or enzyme inhibition activity.

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F. Further Reacti ns:

Piperazinedione mixtures derived from amines #1 and #2 are deprotected with hydrogen bromide in acetic acid as described in example 4, and can be acylated or guanidinated as described above or tested as free amines. All sublots were divided in half and reduced with lithium aluminum hydride as described in example 11.

Example 11. General Procedure for LiAlH₄ Reduction of Mixtures of 10 Piperazinediones from Example 10.

The piperazinediones obtained from cyclization reactions or following protecting group removal, as residues in 5 mL flasks, are carefully treated with 1 M LiAlH4 solution (20 equivalents) and attached to the multiple reaction device. In these instances, a plain glass column replaces the gas dispersion tube and positive nitrogen pressure is provided by plastic tubing attached to the glass column. External cooling of the condensor tubing is provided by packing the exposed glass and plastic tubing in dry ice. The 5 mL flasks are then placed in an 80 ° C oil bath and the reactions refluxed 16 hours. The reactions are cooled to room temperature and poured into ice. The flasks are rinsed well sequentially with dichloromethane, methanol and water. The volatile organic solvents and water are removed on a speed-vac and the residue is extracted with acetonitrile (1.5 mL), evaporated and assayed or submitted to further modification such as acylation or guanidination.

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Example 12. Bradykinin 2 Guinea Pig Ileum Receptor Binding Assay.

Guinea pig ileum membranes were preparation by Analytical Biological Services, Inc. Briefly, ilea were finely chopped and combined with TES buffer (25 mM, pH 6.8) containing 1 mM 1, 10-phenanthroline, 5 ug/ml soybean trypsin inhibitor, 100 ug/ml bacitracin, 1 mM benzamidine, and 100 uM phenylmethylsulfonyl fluoride. This mixture was homogenized in a Brinkman PT-20 Polytron (setting 7, 4 x 20 sec intervals) and subjected to differential centrifugation (1000 x g, 4^{0} C, 10 min). The pellet was discarded and the supernatant centrifuged for 15 min at 43,000 x g at 4^{0} C. The pellet was washed twice by resuspending in the buffer described above and

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centrifuging as above. The pellet was resuspended and stored at -70⁰C until use.

Guinea pig ileum membrane solution were incubated with ³H-bradykinin (final concentration 0.3 nM) with or without test compounds in assay buffer (25 mM TES, pH 6.8 containing 1 mM 1,10 phenanthroline, 1 mM dithiothreitol, 2 uM captopril, 140 ug/ml bacitracin, and 0.1 % bovine serum albumin, BSA), at room temperature, for 45 minutes, at a final volume of 315 ul. All test compound dilutions were done in triplicate. Assays were harvested by quick filtration in a Tomtec Harvester 96, with ice-cold wash buffer consisting of 10 mM Tris/HCl, pH 7.5, 100 mM NaCl, 02% BSA, onto Wallec printed glassfiber Filtermat "B", which had been pre-soaked with 0.1% polyethyleneimine (PEI) and previously airdried. Filtermats were counted in 9.5 ml Wallec Beta-Plate Scint, in Wallec 1450 MicroBeta Counter.

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Example 13. Bradykinin 2 and Bradykinin 1 Human Receptor Clone Binding Assays.

Human bradykinin B₂ receptor was expressed in CHO-K1

20 (ATCC) cells. Preparation of membranes for binding assay was carried out by scraping cells from roller bottles in ice cold PBS and centrifuging at 1000 x g, at 4°C for 15 minutes. The supernatant was discarded and pellet resuspended in Buffer A consisting of 25 mM TES(pH 6.8) with 2uM 1,10-phenanthroline, and centrifuged at 27,000 x g for 15 min. The pellet was washed once using the same buffer and centrifugation parameters. The final pellet was resuspended in Buffer B (Buffer A with 2uM captopril, 140 ug/ml bacitracin, 0.1% BSA), and stored in 1 ml aliquot, frozen at -70°C until needed.

B2 Binding assays were performed by incubating human clone
membrane solution with ³H-bradykinin (final concentration 0.3 nM) with or
without test compounds in assay buffer (Buffer B with 1 mM dithiothreitol),
at room temperature, for 45 minutes, at a final volume of 315 ul. All test
compound dilutions were done in triplicate. Assays were harvested by
quick filtration in a Tomtec Harvester 96, with ice-cold wash buffer
consisting of 10 mM Tris/HCl, pH 7.5, 100 mM NaCl, 02% BSA, onto Wallec
printed glassfiber Filtermat "B", which had been pre-soaked with 0.1% PEI

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and previously air-dried. Filtermats were counted in 9.5 ml Wallec Beta-Plate Scint, in Wallec 1450 MicroBeta Counter.

Human lung fibroblasts IMR-90 cells were obtained from ATCC and propagated in DMEM media in 850 mm roller bottles until confluent. Three hours prior to harvesting, the cells were treated with interleukin 1-beta (200 pg/ml). After three hours, cells were scraped from roller bottles in ice cold PBS and centrifuging at 1000 x g, at 4°C for 15 minutes. The supernatant was discarded and pellet resuspended in Buffer A consisting of 25 mM TES(pH 6.8) with 2 uM 1,10-phenanthroline, and centrifuged at 27,000 x g for 15 min. The pellet was washed once using the same buffer and centrifugation parameters. The final pellet was resuspended in Buffer B (Buffer A with 2 uM captopril, 140 ug/ml bacitracin, 0.1% BSA), and stored in 1 ml aliquot, frozen at -70°C until use.

The human bradykinin B1 binding assays were performed by incubating IMR-90 membrane solution (approx. 150 ug/well) with ³H-des-Arg¹⁰Kallidin, (final concentration 0.5 nM) with or without test compounds in assay buffer (Buffer B with 1 mM dithiothreitol), at room temperature, for 45 minutes, at a final volume of 200 ul. All test compound dilutions were done in triplicate. Assays were harvested by quick filtration in a Tomtec Harvester 96, with ice-cold wash buffer consisting of 10 mM Tris/HCl, pH 7.5, 100 mM NaCl, 0.02% BSA, onto Wallec printed glassfiber Filtermat "B", which had been pre-soaked with 0.1% PEI and previously air-dried. Filtermats were counted in 9.5 ml Wallac Beta-Plate Scint, in Wallace 1450 MicroBeta Counter.

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Example 14. Neurokinin 1 Human Receptor Clone Binding Assay.

The human NK-1 receptor was expressed in CHO-K1 (ATCC) cells. Preparation of membranes for binding assay was carried out by scraping cells from culture flasks in ice cold PBS and centrifuging at 500 x g, at 4°C for 10 minutes. The supernatant was discarded and pellet resuspended in 50 mM Tris/HCl, pH 7.4 with 1 mM 1,10 phenanthroline, and centrifuged at 27,000 x g for 15 min. The pellet was washed once using 20 mM HEPES, pH 7.4, 5 mM MgCl₂, 30 mM KCl, 0.02% BSA and 100 uM thiorphan and centrifuged as above. The final pellet was resuspended in the wash buffer (above) and stored in 1 ml aliquot, frozen at -70°C until needed.

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Human NK1 receptor binding assays were performed by incubating human clone membrane solution (50 ug/well in 125 ul) with ³H-substance P (final concentration 1.0 nM) with or without test compounds in assay buffer (Buffer B with 1 mM dithiothreitol), at room temperature, for 30 minutes, at a final volume of 315 ul. All test compound dilutions were done in triplicate. Assays were harvested by quick filtration in a Tomtec Harvester 96, with ice-cold wash buffer consisting of 20 mM Tris/HCl, pH 7.5, 120 mM NaCl, 5 mM KCl and 0.02% BSA, onto Wallec printed glassfiber Filtermat "B", which had been pre-soaked with 0.1% PEI and previously air-dried. Filtermats were counted in 9.5 ml Wallec Beta-Plate Scint, in Wallec 1450 MicroBeta Counter.

Example 15. Neurokinin 2 Human Receptor Clone Binding Assay.

The cell membrane preparation and receptor binding procedure are identical to that described above for the NK-1 receptor (Example 14) except that ³H-neurokinin A was used as the ligand at 2.0 nM.

20 Example 16. Mu-Opioid Human Receptor Clone Binding Assay.

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The human mu-opioid receptor was expressed in CHO-K1 (ATCC) cells. Preparation of membranes for binding assay was carried out by scraping cells from culture flasks in ice cold PBS and centrifuging at $500 \times g$, at 4° C for 10 minutes. The supernatant was discarded and pellet resuspended in 10 mM Tris/HCl, pH 7.4 with 0.32 M sucrose and centrifuged for 30 min at 40° C at $27,000 \times g$. The supernatant was discarded and the pellet resuspended in the wash buffer (above) and stored in 1 ml aliquot, frozen at -70°C until needed.

Human mu-opioid binding assays were performed by

incubating human clone membrane solution (50 ug/well in 125 ul) with ³HDAMGO ([D-Ala², N-Me-Phe⁴, Gly⁵-ol]enkephalin; final concentration 5.0
nM) with or without test compounds in assay buffer (Buffer B with 1 mM
dithiothreitol), at room temperature, for 60 minutes, at a final volume of 315
ul. All test compound dilutions were done in triplicate. Assays were

harvested by quick filtration in a Tomtec Harvester 96, with ice-cold wash
buffer consisting of 50 mM Tris/HCl, pH 7.4 onto Wallec printed glassfiber

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Filtermat "B", which had been pre-soaked with 0.1% PEI and previously airdried. Filtermats were counted in 9.5 ml Wallec Beta-Plate Scint, in Wallec 1450 MicroBeta Counter.

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Example 17. k-Opioid Guinea Pig Brain Receptor Binding Assay.

Preparation of guinea pig brain membrane for Kappa binding assay was carried out by homogenizing tissue with 5 second bursts of tissue homogenizer, until smooth, in ice-cold assay buffer consisting of 10 mM 10 Tris/HCl, pH 7.4 with 0.32 M Sucrose. The homogenate was centrifuged at 1,500 x g, for 10 minutes at 4°C. The pellet was discarded and supernatant was centrifuged at 30,000 x g, for 30 min at 4°C. Supernatant was discarded and pellet resuspended in fresh assay buffer, and in 1 ml aliquot, frozen at -70°C until needed. Protein concentration was determined by Bradford analysis.

k-Opioid receptro binding assays were performed by incubating guinea pig brain membrane solution (50 ug/well in 125 ul final concentration) with ³H-U69,593 (final concentration 5.0 nM, New England Nuclear) with or without test compounds in assay buffer, at room temperature, for 60 minutes, at a final volume of 315 ul. All test compound dilutions were done in triplicate. To block mu receptor sites, 0.1 uM solution of unlabeled DAMGO, in assay buffer, was added to all wells. Assays were harvested by quick filtration in a Tomtec Harvester 96, with ice-cold wash buffer consisting of 50 mM Tris/HCl, pH 7.4, onto Wallec printed glassfiber Filtermat "B", which had been pre-soaked with 0.1% PEI and previously air-dried. Filtermats were counted in 9.5 ml Wallec Beta-Plate Scint, in Wallec 1450 MicroBeta Counter.

30 Example 18. Bradykinin Induced Calcium Flux in Human WI38 Fibroblasts.

A fluorescent indicator (Fura-2) method was used to measure cytosolic Ca(+2) (A. Pilyavskaya et al. FASEB J. 9, 6, A1371, 1995). Human lung fibroblasts are loaded with Fura-2 and, after incubation with BK in the absence and presence of antagonists, the cells are analyzed in a Perkin-Elmer LS50B spectrofluorometer using excitation wavelengths of 340 & 380

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nm and an emission wavelength of 510 nm. Cells are maintained in suspension during the measurements.

fethod B.	R -(CH ₂), _N NH ₂	R.S -(CH ₂), -N	Mass Spec. CP (M+H)	2006						2051					2076									2104
ng to N			Ma	18	343	386	372	380	406	372	386	380	393	372	386	380	393	394	386	386	394	425	412	440
red Accordir	, R6C =	R6F =	HPLC a	20.0	17.8	19.6	17.3	18.1	21.2	16.5	19.7	19.5	19.5	17.0	17.3	17.8	19.1	18.3	18.7	17.6	18.0	17-18	20.3/22.1	17.4/18.0
nediones Prepa	S -(CH ₂), - N H	R,S -(CH ₂),—NH ₂	<u>R</u> 6	R6A	R6D	R6A	R6B	R6B	R6B	R6B	R6B	R6B	R6B	R6C	R6C	R6C	R6C	R68	R6B	R6C	R6C	R6E	R6F	R6F
and Piperazii	R6B = 8	R6E = R.	\mathbb{R}^{Z}	-CH ₂ -C ₆ H ₁₁	-CH2-C ₆ H ₁₁	-CH2-C ₆ H ₁₁	-C ₆ H ₁₁	benzyl	2-indanyl	$-C_6H_{11}$	-CH ₂ -C ₆ H ₁₁	benzyl	-CH2CH2Ph	$-C_6H_{11}$	-CH2-C ₆ H ₁₁	benzyl	-CH2CH2Ph	-CH2CH2Ph	$-CH_2-C_6H_{11}$	CH_2 - C_6H_{11}	-CH2CH2Ph	3,4 Di-MeOBn	-CH2-C ₆ H ₁₁	3,4 Di-MeOBn
Specific Piperazines and Piperazinediones Prepared According to Method B.	S ·(CH ₂) ₃ -N NH ₂	S -{CH ₂ } ₃ -NH ₂	\mathbb{R}^5	R-benzyl	R-benzyl	R-benzyl	R-benzyl	R-benzyl	R-benzyl	S-benzyl	S-benzyl	S-benzyl	S-benzyl	S-benzyl	S-benzy!	S-benzyl	Spenzyl	R-benzyl	R-benzyl	R-benzyl	R-benzyl	R,S-2-indanyl	R,S-2-indanyl	R,S,-benzyl
19 - 39 .	R6A =	R6D =	×I	0	H2	H2	H ₂	H_2	H_2	H_2	7	H_2	H ₂	H_2	H_2	H_2	H_2	H_2	H_2	H_2	H_2	H_2	H_2	H_2
Examples 19	_	_	Example	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39

	24 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	. Specific Piper	azines and Pi	perazihedione	s Prepared	Specific Piperazines and Piperazihediones Prepared According to Method A.	thod A.
	S13	(CH ₂) ₄ -N H	S14 n	·(CH ₂) ₅ N	<u></u>	(CH ₂), -N-1, -N-	
	S2 ==	-(CH ₂) ₆ -N H	S12 ==				
#JO	Example	×	R1	N	<u>K3</u>	HPLC	MS
2114	40	H2	R,S-benzyl	-CH2-C ₆ H ₁₁	S13	19.9	386 386
2115	41	H2	R,S-benzyl	-CH2-C6H11	S14	19.8	400
2116	42	H2	R,S-benzyl	-CH ₂ -C ₆ H ₁₁	S2	19.5	414
2117	43	H2	R,S-benzyl	-CH2-C ₆ H ₁₁	S1	20.2	428
2151	44	0	S-benzyl	I	S1	16.7	360
2371	45	0	S-3-indole	$-CH_2CH_2Ph$	S12	24.1	521
2372	46	0	R-3-indole	-CH ₂ CH ₂ Ph	S12	24.1	521
2373	47	0	S-benzyl	-CH2CH2Ph	S12	24.0	482
2374	48	0	R-benzyl	-CH2CH2Ph	S12	24.0	482
2399	49	0	S-CH2-C6H11	-CH2CH2Ph	S12	29.0	488
2400	20	0	R -CH2-C6H11	-CH2CH2Ph	S12	29.0	488
2425	51	0	S-CH ₂ -2-Nph	$-CH_2CH_2Ph$	S12	27.3	532
2426	52	0	R -CH2-2-Nph	-CH2CH2Ph	S12	27.5	532
2427	23	0	x	-CH2CH2Ph	S12	18.4	392
2428	54	0	R,S-2-indanyl	-CH2CH2Ph	S12	26.5	508
2429	55	0	S-CH ₂ -2-Nph	$-CH_2C_6H_{11}$	S12	29.1	524
2430	56	0	R -CH2-2-Nph	$-CH_2C_6H_{11}$	S12	29.2	524
2431	57	0	R,S-2-indanyl	-CH ₂ C ₆ H ₁₁	S12	29.1	200
2456	28	0	S-CH2-C6H11	$CH_2C_6H_{11}$	S12	31.2	480
2457	59	0	R -CH2-C6H11	$CH_2C_6H_{11}$	S12	31.3	480
2458	09	0	R,S-2-indanyl	CH2-2-Nph	S2	28.1	512
2459	61	0	R,S-2-indanyl	4'-biphenyl	S1	31.0	552

Examples 62 through 437. Combinatorial Mixtures of Piperazindiones and Piperazines Prepared According to Methods 4,6, 10, 11.

$$R^{2} = M = -CH_{2} \bigcirc \qquad -CH_$$

$$S1 = \cdot (CH_2)_7 - N \\ NH_2 \\ S1R = \cdot (CH_2)_7 - NH_2 \\ S2R = \cdot (CH_2)_6 - NH_2 \\ S2 = \cdot (CH_2)_6 - NH - Cbz \\ S3 = \cdot (CH_2)_7 - N \\ NH_2 \\ S4 = \cdot (CH_2)_6 - NH_2 \\ S5 = \cdot (CH_2)_6 - NH_2 \\ S5 = \cdot (CH_2)_6 - NH_2 \\ NH_2 \\ S6R = \cdot (CH_2)_6 - NH_2 \\ NH_2 \\ S8R = \cdot (CH_2)_6 - NH_2 \\ NH_2 \\ S8R = \cdot (CH_2)_6 - NH_2 \\ NH_2 \\ S8R = \cdot (CH_2)_6 - NH_2 \\ NH_2$$

77 78 79 80 81 82 83 84 85 86	O H ₂ H ₂ H ₂ H ₂ H ₂ H ₂ H ₂ H ₂	A A A A A A A	M M M M M M M M	\$2Cb2 \$1R \$2R \$6R \$8R \$9R \$10R \$11R \$12 \$1
Example 88 89 90 91 92 93 94 95 96 97 98 99 100 101 102 103 104 105 106 107 108 109 110 111 112 113	X0000000000000000000000000000000000000	R1 BBBBBBBBBBBBBBBBBBBBBBBBBBBBBBBBBBBB	R2 M M M M M M M M M M M M M M M M M M M	83 S1 S2 S1R S2R S3 S4 S5 S6 S7 S8 S9 S10 S11 S12 S1Cbz S2Cbz S1R S2R S6R S9R S10R S11R S11R S11R S11R S11R S11R S11
Example 114 115 116 117 118 119 120 121 122 123 124 125 126 127 128 129 130 131 132 133 134	X0000000000000000000000000000000000000	R1 000000000000000000000000000000000000	R2 M M M M M M M M M M M M M M M M M M M	83 51 52 51 52 51 53 54 55 56 57 58 59 510 512 51Cbz 52Cbz 51R 52R 52R 56R 58R 59R

135 136 137 138 139	H ₂ H ₂ H ₂ H ₂ H ₂	0000	М М М М	S10R S11R S12 S1 S2
Example 140 141 142 143 144 145 146 147 148 149 150 151 152 153 154 155 156 157 158 159 160 161 162 163 164 165	X0000000000000000000000000000000000000		R2 M M M M M M M M M M M M M M M M M M M	\$3 \$1 \$2 \$1 \$2 \$3 \$3 \$4 \$5 \$5 \$6 \$7 \$8 \$9 \$10 \$11 \$12 \$1Cbz \$2Cbz \$1R \$2R \$6 \$8 \$9 \$10 \$11 \$12 \$10 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$11 \$12 \$12
Example 166 167 168 169 170 171 172 173 174 175 178 179 180 181 182 183 184 185 186 187 188 189 190 191	X0000000000000000000000000000000000000	R1 666666666666666666666666666666666666	R2 M M M M M M M M M M M M M M M M M M M	83 S1 S2 S1R S2R S3 S4 S5 S6 S7 S8 S9 S10 S11 S12 S1Cbz S2Cbz S1R S2R S6R S9R S10R S11R S11R S11R S12 S10R S11R
Example 192 193 194 195 196	X 0 0 0 0	R1 F2 F2 F2 F2 F2 F2	R2 M M M M M	R3 S1 S2 S1R S2R S3

197 198 199 200 201 202 203 204 205 206 207 208 209 210 211 211 212 213 214 215 216 217	00000000000000000000000000000000000000	克克克克克克克克克克克克克克克克克克克克克克克克克克克克克克克克克克克克	M M M M M M M M M M M M M M M M M M M	\$4 \$5 \$5 \$6 \$7 \$8 \$9 \$10 \$12 \$1Cbz \$2Cbz \$1R \$2R \$6R \$9R \$10R \$11R \$11R \$12 \$1 \$15 \$15 \$15 \$15 \$15 \$15 \$15 \$15 \$15
Example 218 219 220 221 222 223 224 225 226 227 228 229 230 231 234 235 234 235 236 237 238 239 240 241 242 243	X0000000000000000000000000000000000000	ដ េកកក្តុក្រុក ខេត្តក្រុកក្រុកក្រុកក្រុកក្រុកក្រុកក្រុកក្	R2 M M M M M M M M M M M M M M M M M M M	R3 S1 S2 S1R S2R S3 S4 S5 S6 S7 S8 S9 S1Obz S1Cbz S1Cbz S1Cbz S1Cbz S1R S2R S6R S9R S10R S11R S11R S11R S11R S11R S11R S11
Example 244 245 246 247 248 249 250 251 252 253 254 255 255 256 257 258 259 260 261	X 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	R1 H H H H H H H H H H H H H H H H H H H	R2 M M M M M M M M M M M M M M M M M M M	R3 S1 S2 S1R S2R S3 S4 S5 S5 S6 S7 S8 S9 S10 S11 S12 S1Cbz S2Cbz S1R S2R

H ₂ H ₂ H ₂ H ₂	н н н	м м м м	S&R S&R S9R S10R
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X0000000000000000000000000000000000000		R2 M M M M M M M M M M M M M M M M M M M	83 S1 S2 S1R S2R S3 S4 S5 S6 S7 S8 S9 S10 S12 S1Cbz S1Cbz S1R S2R S6R S8R S9R S10R S11R S11R S12R S11R S12R S11R S2R S2R S3R S11R S2R S3R S11R S12R S11R S12R S11R S12R S11R S12R S11R S11R S11R S11R S11R S11R S11R S11R S11R S11R S11R S11R S11R S11R S11R S12R S12R S13R S12R S13R S1
H ₂ X O O	R1	R2	<u>R3</u>
	HHHHHHHH X0000000000000000HHHHHHHHHHHHH	H12 H22 H22 X00000000000000000000000000000	H2 H H H M M M M M M M M M M M M M M M M

325 326 327 328 329 330 331 332 333 334 335 336 337 338 339 340 341 342 343 341 342 343 344 345 346 347 348	00000000000000000000000000000000000000	**************************************	M M M M M M M M M M M M M M M M M M M	S1R S2R S3 S4 S5 S6 S7 S8 S9 S10 S11 S12 S1Cbz S2R S2R S6R S9R S10R S11R S12 S1
349 350 351 352 353 354 355 356 357 358 359 360	0000000000	KNH2 KNH2 KNH2 KNH2 KNH2 KNH2 KNH2 KNH2	M M M M M M M M M	S1R S2R S3 S4 S5 S6 S7 S8 S9 S10 S11 S12
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385	0	I _{NH2}	М	S12
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410 411 412 413 414 415 416 417 418 419 420 421	00000000000	ICbz ICbz ICbz ICbz ICbz ICbz ICbz ICbz	M M M M M M M M M	S3 S4 S5 S6 S7 S8 S9 S10 S11 S12 S1Cbz S2Cbz
422 423 424 425 426 427 428 429	H ₂ H ₂ H ₂ H ₂ H ₂ H ₂ H ₂	F1 F1 F1 F1 F1 F1 F1	M M M M M M	\$6R \$8R \$9R \$10R \$11R \$12 \$1 \$2
430 431 432 433 434 435 436 437	H ₂ H ₂ H ₂ H ₂ H ₂ H ₂ H ₂	L L L L L L	M M M M M M	56R 58R 59R 510R 511R 512 51

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Example 438. Description of the Multiple Reaction Device.

The multiple solid phase reaction device consists of polypropylene block clamps which may hold 1 to 12 or more mini-reactors. The mini-reactors consist of 5 mL round bottom flasks (Ace glass model 9592-04) with ground glass joints, and screw adapters, the resin is introduced into a gas dispersion tube (7 mm outside diameter, Ace glass 10 model 9436) with a medium porosity glass fritt. The gas dispersion tube is held in an air tight fit with a ground glass joint screw adapter (Ace model 5028-25). The top of the gas dispersion tube may be fitted to plastic tubing carring nitrogen, solvents or liquid or gaseous reagents. Cooling to allow solvent reflux can be obtained by packing the fixed dispersion tubes in an insulated container of dry-ice. The reaction vessels may be sonicated, swirled on an orbital table, cooled in baths or heated in a high-temperature oil bath. Individual reactors can be opened, heated, or cooled, or left idle, on an individual basis as the need may arise, unlike block reactors described by DeWitt et al.

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WE CLAIM:

- 1. A method of synthesizing 1,4-piperazines and derivatives thereof comprising simultaneously cyclizing and cleaving an optionally substituted glycyl residue or dipeptoid from a polymeric support.
- 2. The method according to claim 1 wherein said cyclization and cleavage is accomplished via heating in a non-reactive solvent in the presence of a catalyst.
- 3. The method according to claim 1 wherein said solvent is 2-butanol and said catalyst is acetic acid.
- The method of synthesizing 1,4-piperazine-2,5-diones
 according to claim 1 further comprising, prior to cylization and cleavage,
 attaching an N-protected α-amino acid to a polymeric
 support;

deprotecting said α -amino acid; reductively alkylating said α -amino acid with an aldehyde or ketone; bromoacetylating resulting secondary amine; and

reacting resulting bromoacetyl compound with a primary amine.

5. The method according to claim 4 wherein said α-amino acid is α-carbon substituted with benzyl, substituted benzyl, indole, methylnaphthyl, cyclohexyl, methylcyclohexyl, indanyl, (C1-C5) alkyloxyaryl, aminoalkyl (C1-C8), guanidino alkyl (C1-C8), alkylcarboxylic, hydroxyalkyl (C1-C5), alkylthioalkyl, (C1-C3) alkylbenzamidio or hydrogen.

- 6. The method according to claim 4 wherein said α -amino acid is selected from from L- or D-Phenylanine, D- or L-Tryptophan, D- or l-Cyclohexylalanine, D- or L- β -2-Naphthylalanine, and D- or L-Indanylglycine.
- 7. The method according to claim 4 wherein said aldehyde is of the formula R1C(O)H, wherein R1 is cyclohexyl, methylcyclohexyl, benzyl, substituted benzyl, phenethyl, biphenyl, benzoyl, hydrogen, methylnapthyl, indanyl, guanidinoalkyl (C2-C8), aminoalkyl (C2-C8), arylalkylcarbamate with alkyl chains of C2 to C8, alkanoyl (C1-C12) or alkyl (C1-C12).
- 8. The method according to claim 3 wherein said aldehyde or ketone group is selected from cyclohexanone, cyclohexanone carboxaldehyde, benzaldehyde, phenylacetalaldehyde, 4'-biphenylcarboxaldehyde, 2-Napthaldehyde, 2-indanone-, 2-(N-Cbz)-ethanal, 6-(N-Cbz)-hexanal and 12-(N-Cbz)-dodecanal.
- 9. The method according to claim 4 wherein said primary amine is of the formula H2NR3 where R3 is selected from cyclohexyl, methylcyclohexyl, benzyl, substituted benzyl, phenethyl, biphenyl, benzoyl, hydrogen, methylnapthyl, indanyl, guanidinoalkyl (C2-C8), aminoalkyl (C2-C8), arylalkylcarbamates with alkyl chains of C2 to C8, alkanoyl (C1-C12), alkyl (C1-C12), (C2-C12)-alkylguanidinoalkyl (-C2-C8), benzamidino, (C1-C6) alkylbenzamidino, (C1-C6) alkylquinuclidine, (C1-C6) alkylpiperazine, C1 to C6 alkyl N-substituted piperazine, (C1-C6) alkylamino piperidine, (C1-C6) alkylcarboxyamidobenzamidino, (C1-C6) alkylcarboxamidoquinuclidine, (C1-C6) alkylcarboxamidopiperazine, C1 to C6 alkylcarboxyl N-substituted piperazine, (C1-C6) alkylcarboxamidino piperidine and (C1-C6) alkylcarboxamidino N-substituted piperidine.

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The method according to claim 4 wherein said primary amine 10. is selected from

- 11. The method of synthesizing piperazines according to claim 1 further comprising reducing resulting 1,4-piperazine-2,5-dione.
- 12. The method according to claim 10 wherein said 1,4-piperazine-2,5-dione is reduced using diborane or lithium aluminum hydride.
- 13. The method of synthesizing 1,4-piperazine-2,5-diones according to claim 1 further comprising, prior to cyclization and cleavage, attaching an N-protected α -amino acid to a polymeric support; deprotecting said α -amino acid; coupling second N-protected α -amino acid residue to attached α -amino acid; deprotecting said residue; and

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reductively alkylating resulting dipeptoid with an aldehyde or ketone group.

- 14. The method according to claim 13 wherein said first α-amino acid is α-carbon substituted with benzyl, substituted benzyl, indole, methylnaphthyl, cyclohexyl, methylcyclohexyl, indanyl, (C1-C5) alkyloxyaryl, aminoalkyl (C1-C8), guanidino alkyl (C1-C8), alkylcarboxylic, hydroxyalkyl (C1-C5), alkylthioalkyl, (C1-C3) alkylbenzamidio or hydrogen.
- 15. The method according to claim 15 wherein said second α-amino acid is α-carbon substituted with cyclohexyl, methylcyclohexyl, benzyl, substituted benzyl, phenethyl, biphenyl, benzoyl, hydrogen, methylnapthyl, indanyl, guanidinoalkyl (C2-C8), aminoalkyl (C2-C8), arylalkylcarbamates with alkyl chains of C2 to C8, alkanoyl (C1-C12), alkyl (C1-C12), (C2-C12)-alkylguanidinoalkyl (-C2-C8), benzamidino, (C1-C6) alkylbenzamidino, (C1-C6) alkylquinuclidine, (C1-C6) alkylpiperazine, C1 to C6 alkyl N-substituted piperazine, (C1-C6) alkylamino piperidine, (C1-C6) alkylcarboxyamidobenzamidino, (C1-C6) alkylcarboxamidoquinuclidine, (C1-C6) alkylcarboxamidopiperazine, C1 to C6 alkylcarboxyl N-substituted piperazine, (C1-C6) alkylcarboxamidino piperidine or (C1-C6) alkylcarboxamidino N-substituted piperidine.
- 16. The method according to claim 13 wherein said α -amino acids are independently selected from L- or D-Phenylanine, D- or L-Tryptophan, D- or l- Cyclohexylalanine, D- or L- β -2-Naphthylalanine, and D- or L-Indanylglycine.
- 17. The method according to claim 13 wherein said aldehyde is of the formula R7C(O)H, where R7 is cyclohexyl, methylcyclohexyl, benzyl, substituted benzyl, phenethyl, biphenyl, benzoyl, hydrogen, methylnapthyl,

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indanyl, guanidinoalkyl (C2-C8), aminoalkyl (C2-C8), arylalkylcarbamate with alkyl chains of C2 to C8, alkanoyl (C1-C12), or alkyl (C1-C12).

- 18. The method according to claim 13 wherein said aldehyde or ketone is selected from cyclohexanone, cyclohexanone carboxaldehyde, benzaldehyde, phenylacetalaldehyde, 4'-biphenylcarboxaldehyde, 2-Napthaldehyde, 2-indanone-, 2-(N-Cbz)-ethanal, 6-(N-Cbz)-hexanal, and 12-(N-Cbz)-dodecanal.
- 19. A method of synthesizing [a combinatorial library of] optionally substituted piperazine derivatives comprising simultaneously cyclizing and cleaving optionally substituted glycyl residues from a polymeric support.
- 20. A method of synthesizing (W)(X)(Z) N-substituted1,4-piperazine-2,5-dione derivatives according to claim 19, further comprising, prior to cyclization and cleavage,

attaching W N-protected α -amino acids to W polymeric support resins, where W is the number of chosen α -amino acids;

mixing 1/W of each resin into W mixtures, each mixture having an equivalent amount of each resin;

dividing said mixtures into X aliquots, where X is the number of chosen aldehyde or ketone groups;

deprotecting said α -amino acids;

reductively alkylating said α -amino acids in each (X)(W) aliquot with a different aldehyde or ketone group, where (X)(W) is the total number of aliquots;

bromoacetylating resulting secondary amines;

dividing each (X)(W) aliquots into Z aliquots, where Z is the number of chosen primary amines; and

reacting each (X)(W)(Z) aliquot with a different primary amine, where (X)(W)(Z) is the total number of aliquots reacted with amines.

- 21. The method according to claim 20 further comprising dividing said (X)(W)(Z) aliquots of derivatized piperazinediones into two equal portions, Z1 and Z2; and reducing derivatized piperazinediones of Z1 aliquots to piperazine derivatives.
- 22. The method according to claim 20 wherein α -amino acid is α -carbon substituted with benzyl, substituted benzyl, indole, methylnaphthyl, cyclohexyl, methylcyclohexyl, indanyl, (C1-C5) alkyloxyaryl, aminoalkyl (C1-C8), guanidino alkyl (C1-C8), alkylcarboxylic, hydroxyalkyl (C1-C5), alkylthioalkyl, (C1-C3) alkylbenzamidio or hydrogen.
- 23. The method according to claim 20 wherein said aldehyde is of the formula R1C(O)H, wherein R1 is cyclohexyl, methylcyclohexyl, benzyl, substituted benzyl, phenethyl, biphenyl, benzoyl, hydrogen, methylnapthyl, indanyl, guanidinoalkyl (C2-C8), aminoalkyl (C2-C8), arylalkylcarbamate with alkyl chains of C2 to C8, alkanoyl (C1-C12) or alkyl (C1-C12).
- 24. The method according to claim 20 wherein said primary amine is of the formula H2NR3 where R3 is selected from cyclohexyl, methylcyclohexyl, benzyl, substituted benzyl, phenethyl, biphenyl, benzoyl, hydrogen, methylnapthyl, indanyl, guanidinoalkyl (C2-C8), aminoalkyl (C2-C8), arylalkylcarbamates with alkyl chains of C2 to C8, alkanoyl (C1-C12), alkyl (C1-C12), (C2-C12)-alkylguanidinoalkyl (-C2-C8), benzamidino, (C1-C6) alkylbenzamidino, (C1-C6) alkylquinuclidine, (C1-C6) alkylpiperazine, C1 to C6 alkyl N-substituted piperazine, (C1-C6) alkylamino piperidine, (C1-C6) alkylcarboxyamidobenzamidino, (C1-C6) alkylcarboxamidoquinuclidine, (C1-C6) alkylcarboxamidopiperazine, C1 to C6 alkylcarboxyl N-substituted piperazine, (C1-C6) alkylcarboxamidino piperidine and (C1-C6) alkylcarboxamidino N-substituted piperidine and (C1-C6) alkylcarboxamidino N-substituted piperidine.

- 25. The method according to claim 20, wherein W is 10;X is 10; andZ is 12.
- 26. The method according to claim 25, wherein said α-amino acids are L- and D-Phenylanine, D- and L-Tryptophan, D- and L-Cyclohexylalanine, D- and L-β-2-Naphthylalanine, and D- and L-Indanylglycine; said aldehydes and ketones are cyclohexanone, cyclohexanone carboxaldehyde, benzaldehyde, phenylacetalaldehyde, 4'-biphenylcarboxaldehyde, 2-Napthaldehyde, 2-indanone-, 2-(N-Cbz)-ethanal, 6-(N-Cbz)-hexanal and 12-(N-Cbz)-dodecanal; and said primary amines are

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- 27. The N-substituted 1,4-Piperazines made according to claim 21.
- 28. The N-substituted 1,4-Piperazinediones made according to claim 19.

29. A compound of the formula:

$$R^5$$
 N
 R^7
 R^6

wherein R⁵ is benzyl or 2-indanyl; R⁶ is $-(CH_2)_4NHC(=NH)NH_2$; and R⁷ is 2-indanyl, $-CH_2C_6H_{11}$, benzyl, or $-CH_2CH_2Ph$.

- 30. The compound of claim 29 wherein R⁵ is benzyl.
- 31. The compound of claim 30 wherein R⁵ is R-benzyl.
- 32. The compound of claim 31 wherein R^6 is $S (CH_2)_4 NHC (= NH) NH_2$; and R^7 is 2-indanyl or $-CH_2C_6H_{11}$.
- 33. The compound of claim 31 wherein R^6 is $R (CH_2)_4 NHC (= NH) NH_2$; and R^7 is $-CH_2C_6H_{11}$ or $-CH_2CH_2Ph$.
- 34. The compound of claim 30 wherein R⁵ is S-benzyl.
- 35. The compound of claim 34 wherein R⁶ is S -(CH₂)₄NHC(=NH)NH₂; and R⁷ is -CH₂C₆H₁₁, benzyl or -CH₂CH₂Ph.
- 36. The compound of claim 29 wherein R⁵ is 2-indanyl; and R⁷ is -CH₂C₆H₁₁.
- 37. A compound of the formula:

$$R^2$$
 N
 R^3

wherein R^1 is benzyl; R^2 is $-CH_2C_6H_{11}$; and R^3 is $-(CH_2)_7NHC(=NH)NH_2$ or $-(CH_2)_6NHC(=NH)NH_2$.

38. A compound of the formula:

$$\begin{array}{c|c}
R^2 & & & \\
R^1 & & & \\
\end{array}$$

wherein R¹ is -CH₂C₆H₁₁, -CH₂-2-Nph, or 2-indanyl; R² is -CH₂CH₂Ph, -CH₂C₆H₁₁, -CH₂-2-Nph or 4'-biphenyl; and R³ is -(CH₂)₇NHC(=NH)NH₂, -(CH₂)₆NHC(=NH)NH₂ or

39. A compound of claim 38 wherein R³ is

- 40. A compound of claim 39 wherein R² is -CH₂CH₂Ph.
- 41. A compound of claim 40 wherein R¹ is S -CH₂C₆H₁₁.
- 42. A compound of claim 40 wherein R¹ is R or S -CH₂-2-Nph.

- 43. A compound of claim 39 wherein R¹ is 2-indanyl and R² is -CH₂C₆H₁₁.
- 44. A compound of claim 38 wherein R¹ is 2-indanyl and R² is CH₂-2-Nph.
- 45. A compound of claim 44 wherein R^1 is S-2-indanyl and R_3 is $-(CH_2)_6NHC(=NH)NH_2$.

INTERNATIONAL SEARCH REPORT

Inte onal Application No PCT/US 96/14569

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B. FIELD	S SEARCHED		
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Documenta	ation searched other than minimum documentation to the extent the	hat such documents are included in the fi	elds scarched
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varit and n	nailing address of the ISA European Patent Office, P.B. 5818 Patentiaan 2 NL - 2280 HV Rijswijk Tel. (+ 31-70) 340-2040, Tx. 31 651 epo nl,		
	Fax: (+31-70) 340-3016	Luyten, H	

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